HALF-WIDTHS AND LINE SHIFTS FOR TRANSITIONS IN THE v_3 BAND OF METHANE IN THE 2726-3200 CM⁻¹ SPECTRAL REGION FOR ATMOSPHERIC APPLICATIONS





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abstract

Complex Robert-Bonamy calculations of the pressurebroadened half-width and the pressure induced line shift are made for some four thousand transitions in the v_3 band of methane with N_2 , O_2 , and air as the perturbing gases. This work focuses on A and F symmetry transitions in the spectral range 2726 to 3200 cm⁻¹. More work is needed on the intermolecular potential before calculations can be made for the E-symmetry transitions. The calculations are made at 225 and 296K in order to determine the temperature dependence of the half-width. The calculations are compared with measurements. These data are to support remote sensing of the Earth and Titan atmospheres.





Complex Robert-Bonamy Theory

The model chosen is the semiclassical complex formalism of Robert and Bonamy (CRB) for several reasons.

- 1) The formalism is complex valued, yielding halfwidths and line shifts from a single calculation.
- 2) The cumulant expansion incorporated in the RB formalism allows for the inclusion of higher order terms in the S matrix, eliminating an awkward cutoff procedure that characterized earlier theories.
- 3) The intermolecular dynamics are treated more realistically than in earlier theories, i.e. using curved rather than straight line trajectories.
- 4) Connected to item (3) is the incorporation in the CRB theory of a short range (Lennard-Jones 6-12) atom-atom component to the intermolecular potential. This component has been shown to be essential for a proper description of pressure broadening.







Halfwidth and Line Shift in RB theory

In computational form the half-width and line shift are usually expressed in terms of the Liouville scattering matrix

$$(\gamma - i\delta)_{f \leftarrow i} = \frac{n_2}{2 \pi c} \sum_{J_2} \langle J_2 | \rho_2 | J_2 \rangle \int_0^\infty v f(v) dv \int_0^\infty 2\pi b \left[1 - e^{-\{S_1 + \operatorname{Im}(S_2)\}} e^{-\operatorname{Re}(S_2)} \right] db$$

where n_2 is the number density of perturbers and the average is over all trajectories given by impact parameter b and initial relative velocity, v, and initial rotational state, J₂, of the collision partner.





Mean-relative thermal velocity approximation

$$\gamma = \frac{n_2 \overline{v}}{2\pi c} \sum_{J_2} \left\langle J_2 | \rho_2 | J_2 \right\rangle \int_0^\infty 2\pi b \, db \left[1 - \cos \left\{ S_1 + \operatorname{Im}(S_2) \right\} e^{-\operatorname{Re}(S_2)} \right]$$

Terms from from imaginary part cumulant expansion

$$\delta = \frac{n_2 \overline{v}}{2\pi c} \sum_{J_2} \left\langle J_2 | \rho_2 | J_2 \right\rangle \int_0^\infty 2\pi b \, db \left[\sin \left\{ S_1 + \operatorname{Im}(S_2) \right\} e^{-\operatorname{Re}(S_2)} \right]$$





From measurements of the half-width the following conclusions are reached:

The vibrational dependence of the half-width has been found to be of order a few percent for low J lines, but can be of order 20% for high J lines.

The half-widths decrease with J" but there is no simple propensity rule to describe the behavior and the branch dependence is small; of order a few percent.





Devi *et al.* (*J. Mol. Spectrosc.* **157** (1993) 95-111.) have found A, E and F symmetry level transitions have almost the same average pressure-broadening coefficient.

- The dependence on the perturbing molecule/atom show the ratios as being somewhat constant, indicating that predictions for other perturbers can be obtained once the ratio of one transition is known.
- The shifts demonstrate a marked dependence on vibrational band. From the studies made the results indicate that most of the shifts are negative and transition dependent.







Calculations

The 2004 HITRAN database contains some 251 440 methane transitions. Considering only bands of ${}^{12}CH_4$ between 9 and 1.6 µm with the sum of line intensities greater than 10^{-20} cm⁻¹/(molecule·cm⁻²) yields a list of over one hundred thousand transitions.

Here, we focus on the region 2726 to 3200 cm⁻¹, which still contains over twenty five thousand transitions for ${}^{12}CH_4$. The dominant band in this region is the v₃ band centered at 3018 cm⁻¹.





Calculations

In the present work we find convergence of the atom-atom part of the intermolecular potential is obtained at 12th order for transitions involving A-and F-symmetries.

However, for E-symmetry transitions the results show that the calculations are not converged even at 14th order of expansion. Since it is the current limit of the codes, results for E symmetry transitions are not reported here.





Components of the integrand

CH4-N2 (Symmetry-A)



Rc

Components of the integrand

CH4-N2 (Symmetry-F)



Components of the integrand

CH4-N2 (Symmetry-E)



Rc

ine

Real part of the integrand

CH4-N2 for J=8



Rc

Calculations

- From the remaining A- and F-symmetry transitions we were able to make calculations for
 - 524 A-symmetry transitions and
 - 3596 F-symmetry transitions;

4120 v_3 transitions total.





Preliminary Calculations

Combination rules \rightarrow atom-atom parameters with large uncertainty

An initial calculation of the 52 A- and F-symmetry transitions for which Pine (*J. Chem. Phys.* **97** (1992) 773-785.) made N_2 - and O_2 -broadening measurements was done yielding -8.8 and 15.0 percent difference, respectively.





Preliminary Calculations

To adjust the atom-atom parameters six transitions (1 Aand 5 F-species) and five F-symmetry transitions were chosen for N_2 - and O_2 -broadening, respectively.

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Results of fit gives:average percent differenceN_20.15O_2-2.1
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Atom-atom coefficients

Pair	ε/k _B (K)		σ /(Å)		
	initial	final	initial	final	
H-N	20.50	20.50	3.00	2.85	
C-N	34.30	32.59	3.45	3.45	
H-O	24.15	24.15	2.84	2.70	
C-O	40.43	38.01	3.28	3.02	

All other molecular constants are the best literature values.





Perturber	γ_{\min}^{*}	γ_{max}^{*}	δ_{\min}^{*}	δ_{\max}^{*}	N _{min}	<i>n</i> _{max}
N_2	0.0169	0.0657	-0.0208	0.0009	-0.546	0.759
O ₂	0.0251	0.0606	-0.0194	-0.0004	-0.047	0.752
air	0.0186	0.0646	-0.0205	0.0006	-0.398	0.757

Minimum and maximum values of the half-width, line shift, and temperature dependence of the half-width (temperature exponent) for N_2 -, O_2 -, and air-broadening of methane







Calculated air-broadened half-widths of CH₄ versus J".







Air-induced line shifts for v_3 transitions of methane versus J".







Temperature exponents for air-broadened half-widths of v_3 transitions of CH₄ versus J".



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m

Measured N₂-broadened half-widths with error bars and the CRB calculated half-widths (\star symbol) are plotted versus *m*. The measurements are Pine 1992 A-species solid circle, F-species open circle, Pine 1997 A-species solid square, F-species open square, Benner *et al.* 1993 A-species solid delta symbol, F-species open delta symbol, and Mondelain *et al.* 2007 A-species solid star symbol, F-species open star symbol







Percent difference between the CRB calculations and the N_2 - and the O_2 -broadening data of Pine (1992 and 1997) versus *m*.



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Predicting half-widths

9 and 1.6 μ m with S_v >10⁻²⁰ cm⁻¹/(molecule·cm⁻²) > one hundred thousand transitions





Predicting half-widths

Is it possible to predict half-widths for A-/F-species transitions from F-/A-species transitions with the same quantum numbers?

The pairs of lines are chosen such that the upper and lower rotational quantum numbers are the same, the A_1 symmetry is paired with the F_1 symmetry and the A_2 symmetry is paired with the F_2 symmetry, the lower order index values are the same.

Because the allowed order index for A- and F-symmetry states are quite different, the pairs were created from the minimum N'and for the maximum N' of each symmetry species.

For example the line

 $J'A_1N'_{\min} \leftarrow J''A_2N''$ is paired with $J'F_1N'_{\min} \leftarrow J''F_2N''$.







Ratios of A-symmetry transitions to corresponding F-symmetry transition versus J".





Predicting half-widths

Is it possible to predict half-widths for transitions of the same overall symmetry species, i.e. $C''=A_1$ to predict $C''=A_2$, etc?

Transitions are chosen such that

 $J'A_1N' \leftarrow J''A_2N''$ is paired with $J'A_2N' \leftarrow J''A_1N''$

with a similar pairing for F-symmetry







m

Ratios of transitions with upper and lower symmetries interchanged versus *m*.





Predicting half-widths

Is it possible to predict half-widths for transitions of the same J and symmetry species from lines with different upper state counting (order) indices, i.e. N'=5 to predict N'=1, etc?

To test the effect of the upper states on the calculation, we randomly looked through the N_2 -broadening data file and selected the 15 F1 1 lower state. This lower state is involved in transitions to 32 upper states; eleven to 14 F2 states, ten 15 F2 states, and eleven 16 F2 states.







order index

 N_2 -broadened half-widths for v_3 transitions of methane with lower state 15 F1 1.





Predicting half-widths

Is it possible to predict half-widths for methane transitions by scaling as a function of perturbing molecule?

Using the CRB calculated N₂-, O₂-, and airbroadened files, the ratios $\gamma(O_2)/\gamma(N_2)$ $\gamma(air)/\gamma(N_2)$ were determined and plotted.







Ratios of O_2 -broadened half-width to corresponding N_2 -broadened half-width for v_3 transitions of CH₄ versus *m*.





Conclusion

➤ Calculations of the pressure-broadened half-width, the pressure induced line shift, and the temperature dependence of the half-width have been made for nitrogen-, oxygen-, and air-broadening of v_3 transitions of CH₄ in the 2726 to 3200 cm⁻¹ region of the spectrum. The calculations were made at two temperatures, 225 and 296 K, and for rotational states with J≤28.

Because of limitations in the current codes only A- and F symmetry transitions were considered.

➤ Calculations were made for 4120 A- and F- symmetry transitions in this region. The final list of transitions is comprised of 524 Asymmetry transitions and 3596 F-symmetry transitions. The calculations did not take into account any line coupling effects.





➢From the calculations the temperature dependence of the nitrogen-, oxygen-, and air-broadened half-widths were determined.

>Calculations of γ compared with Pine's measurements give reasonable percent differences and standard deviations.

≻However, Mondelain *et al.* have demonstrated that retrievals of methane volume mixing ratios from balloon spectra made using slightly different line shape parameters show noticeable effects. Thus our calculations are not yet at the level desired by all researchers for terrestrial atmospheric applications but certainly are an improvement of the estimated values.





The accuracies needed for Titan remote sensing are less stringent (5-10%) and the CH_4 -N₂ calculations presented here meet that criterion.

≻The comparisons for the line shifts were not as good.

➤This may be due in part to the magnitude of the parameter as well as the uncertainty in the parameters used in the calculation (vibrational-dependence of the polarizability of methane and the atom-atom coefficients).







Future work

>While the agreement with the measured data of Pine gives good statistics (see above) the differences vary between roughly $\pm 10\%$ with a few stray points.

> The fact that the calculations cannot yet be done for E-symmetry transitions is a concern.

➢ Investigate the form of intermolecular potential used in the calculations and the parameters used with emphasis on reducing the spread in the agreement with Pine and being able to compute E-symmetry transitions as well.

>Wavefunctions for v_4 states available, which will allow the consideration of vibration dependence of the half-widths and line shifts (v_3 vs. v_4).

Extend calculations to other broadening species, i.e. H_2 and He.





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